

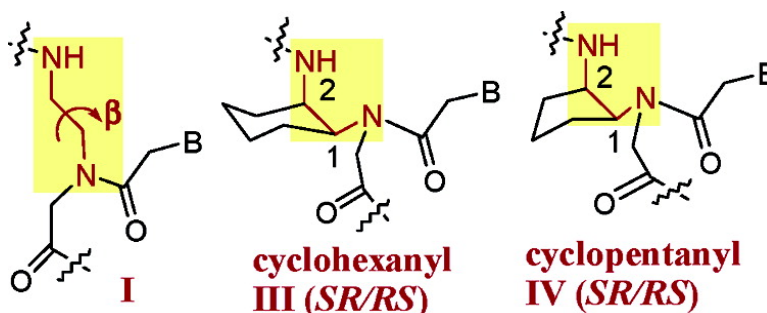
Communication

(*SR/RS*)-Cyclohexanyl PNAs: Conformationally Preorganized PNA Analogues with Unprecedented Preference for Duplex Formation with RNA

T. Govindaraju, Vaijayanti A. Kumar, and Krishna N. Ganesh

J. Am. Chem. Soc., **2005**, 127 (12), 4144-4145 • DOI: 10.1021/ja044142v • Publication Date (Web): 04 March 2005

Downloaded from <http://pubs.acs.org> on March 24, 2009



More About This Article

Additional resources and features associated with this article are available within the HTML version:

- Supporting Information
- Links to the 2 articles that cite this article, as of the time of this article download
- Access to high resolution figures
- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article

[View the Full Text HTML](#)

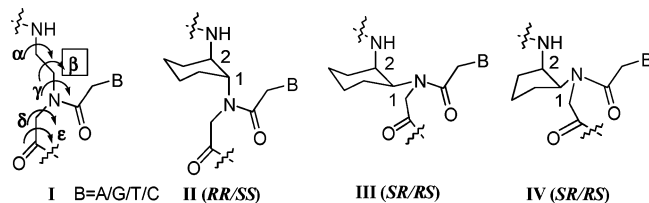
(*SR/RS*)-Cyclohexanyl PNAs: Conformationally Preorganized PNA Analogues with Unprecedented Preference for Duplex Formation with RNA

T. Govindaraju, Vaijayanti A. Kumar,* and Krishna N. Ganesh*

Organic Chemistry Division (Synthesis), National Chemical Laboratory, Pune 411008, India

Received September 27, 2004; E-mail: kng@ems.ncl.res.in; vakumar@dalton.ncl.res.in

Peptide nucleic acid (PNA **I**) is a neutral DNA mimic derived by replacement of the anionic phosphate–sugar backbone with *N*-(1-aminoethyl)glycine units carrying A, G, C, and T nucleobases via methylene carbonyl linkages.¹ Polypyrimidine PNA forms PNA₂:DNA/RNA triplex, and mixed purine–pyrimidine PNA forms a duplex with complementary DNA/RNA strands.² Due to the high binding strength of derived complexes and biostability, PNA has a great potential for development as an antisense agent.³ However, it has drawbacks, such as low cell penetration, similar affinity to DNA/RNA for complementary binding, and unlike other DNA mimics, it forms hybrids in both parallel (*p*) and antiparallel (*ap*) orientations. These would lead to undesired effects on the specificity for target sequences. Some of the PNA deficiencies have been addressed by rational chemical modifications.⁴ However, a sufficiently important problem from an application perspective that has not been adequately dealt with is that of nondiscrimination of identical DNA and RNA sequences.



A comparison of the available X-ray and NMR structural data (Table 1) reveals that the preferred values for the backbone dihedral angle β in PNA₂:DNA triplex⁵ and PNA:RNA⁶ duplex are in the range of 60–70°, while that for PNA:DNA duplex is about 140°.⁷ It occurred to us that it may be possible to rationally impart DNA/RNA duplex binding selectivity by tuning the dihedral angle β of the flexible ethylene part of the PNA backbone (**I**) through suitable chemical modifications. It was reported in the literature that 1,2-*trans*-(*RR/SS*)-disubstituted cyclohexyl PNA (**II**)⁸ in *aeg*-PNA backbone destabilized complexes with both DNA and RNA.⁸ From molecular modeling studies, it was suggested that cyclohexyl 1,2-substituents are in a diaxial disposition, corresponding to a dihedral angle β of $\sim 180^\circ$.⁸ This is incompatible with the geometric requirements to form either PNA:DNA or PNA:RNA complexes (Table 1). Recently, *trans*-(*SS*)-cyclopentane-T in the *aeg*-PNA-T oligomer and mixmer sequence was shown to stabilize the PNA:DNA complex.⁹

In this context, we designed the *cis*-1,2 equatorial-axial (*SR/RS*)-cyclohexyl PNA (**III**) with β of $\sim 65^\circ$, and the derived *aeg*-*ch*-PNA chimera formed both PNA₂:DNA and PNA₂:RNA triplexes.¹⁰ Substituted cyclohexyl ring is inherently rigid, locked up in either of the two chair conformations, and a relatively flexible system would be a cyclopentyl ring,¹¹ endowed with a puckering option. We made a *cis*-(*SR/RS*)-*cp*-PNA (**IV**) monomer, which had the dihedral angle β of $\sim 25^\circ$, as seen from its crystal structure.^{12a} The derived *cp*-PNA-T₈ homooligomer formed highly stable PNA₂:DNA/RNA triplexes, but without much DNA/RNA discrimination.^{12b}

Table 1. Dihedral Angles in PNA and PNA:DNA/RNA Complexes

| | PNA ₂ :DNA ⁵ | PNA:RNA ⁶ | PNA:DNA ⁷ | <i>ch</i> -PNA (1 <i>S</i> ,2 <i>R</i>) ^a | <i>ch</i> -PNA (1 <i>R</i> ,2 <i>S</i>) ^a | <i>cp</i> -PNA (1 <i>S</i> ,2 <i>R</i>) ^a | <i>cp</i> -PNA (1 <i>R</i> ,2 <i>S</i>) ^a |
|----------|------------------------------------|----------------------|----------------------|---|---|---|---|
| α | −103 | 170 | 105 | 128 | −129 | 84 | −84 |
| β | 73 | 67 | 141 | −63 | 66 | −24 | 25 |
| γ | 70 | 79 | 78 | 76 | −78 | 86 | −86 |
| δ | 93 | 84 | 139 | 119 | −119 | 90 | −90 |

^a Values for monomers.^{9,12}

In this paper, we report the remarkable hitherto unseen DNA/RNA duplex discrimination ($\Delta T_m > 30$ –50 °C) when *ch*-PNA-T units are installed into the *aeg*-PNA backbone in a mixed base sequence, in contrast to the *cp*-PNA-T analogues. This strongly validates our design motifs based on tuning the dihedral angle β to achieve DNA/RNA duplex selectivity.

The *N*-protected enantiomeric (1*S*,2*R*) and (1*R*,2*S*) cyclohexyl-T and cyclopentyl-T monomers^{10,12} were individually introduced into *aeg*-PNA mixed decamer **1** at three thymine positions by solid-phase synthesis on MBHA resin functionalized by *L*-lysine linker, according to standard procedures. Cleavage of PNAs from the resin with TFA-TFMSA was followed by purification using RP HPLC to yield the oligomers **1**–**5**, which were characterized by mass spectral data. The T_m values of various PNAs hybridized with cDNA/RNA for parallel and antiparallel binding were determined from temperature-dependent UV absorbance (Figure 1) or CD plots

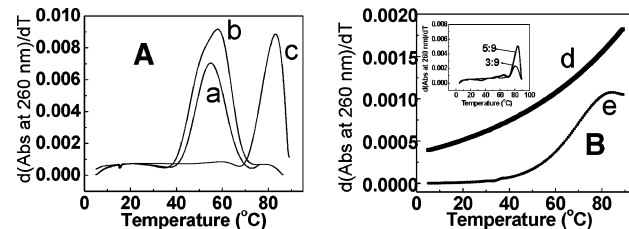


Figure 1. First derivative UV– T_m curves of *ap*-PNA:RNA duplexes (a) **1**:**7**, (b) **2**:**7**, (c) **4**:**7**, (d) **3**:**7**, and (e) **5**:**7**. Inset in B shows derivative UV– T_m of parallel PNA:RNA duplexes **3**:**9** and **5**:**9**.

(Supporting Information) and are summarized in Table 2.

Data in Table 2 show that the reference *aeg*-PNA **1** forms *ap*-duplexes with complementary DNA **6** and RNA **7** with equal stability (entry 1). Compared to the T_m of *aeg*-PNA:DNA duplex, the T_m values of *ch*-PNA:DNA duplexes (entries 2 and 3) are lower ($\Delta T_m = -30$ °C for *SR* and -20 °C for *RS*), and those of *cp*-PNA:DNA duplexes (entries 4 and 5) are higher ($\Delta T_m = +22$ °C for *SR* and $+24$ °C for *RS*). The thermal stability of *ch*-PNA/*cp*-PNA duplexes with complementary DNA/RNA oligonucleotides is stereochemistry dependent with the T_m of *RS* duplexes more than with that of *SR* duplexes. *ch*-PNAs **2** and **3**, having low binding affinity with *ap*-DNA **6**, were ineffective in forming parallel duplexes with DNA **8**.

In the case of PNA:RNA duplexes, with the exception of *SR*-*ch*-PNA **2** (entry 2), other analogues, **3**–**5** (entries 3–5), had a large stabilizing effect ($\Delta T_m > 30$ °C) over that of control *aeg*-

Table 2. T_m ($^{\circ}\text{C}$) of PNA:DNA and PNA:RNA Duplexes^a

| entry | PNA (N \rightarrow C) | DNA 6(8) | RNA 7(9) | $\Delta T_m(\text{RNA}-\text{DNA})_{\text{ap}}$ |
|-------|--|----------------------|----------|---|
| 1 | <i>aeg</i> -GTAGATCACT, 1 | 55 ^b | 55.4 | 0.4 |
| 2 | <i>ch</i> -Gt _{SR} AGAt _{SR} CACT _{SR} , 2 | 25 ^b (nb) | 58 (nb) | 33 |
| 3 | <i>ch</i> -Gt _{RS} AGAt _{RS} CACT _{RS} , 3 | 35 ^b (nb) | >85 (80) | >50 |
| 4 | <i>cp</i> -Gt _{SR} AGAt _{SR} CACT _{SR} , 4 | 77.1 (nb) | 84 (71) | 6.9 |
| 5 | <i>cp</i> -Gt _{RS} AGAt _{RS} CACT _{RS} , 5 | 78.8 (71) | >85 (84) | >6.2 |

^a All values are an average of at least three experiments and are accurate to within ± 0.5 $^{\circ}\text{C}$. Buffer: sodium phosphate (10 mM), pH 7.0, with 100 mM NaCl and 0.1 mM EDTA. ^b Measured by CD to avoid interference from thermal transitions of single-stranded PNAs. ^c A, T, G, and C are *aeg*-PNA bases, T_{SR/RS} *ch*-PNA-T (**2**, **3**); t_{SR/RS} *cp*-PNA-T (**4**, **5**); DNA **6**, 5'-AGTGATCTAC-3'(*ap*); RNA **7**, 5'-AGUGAUCUAC-3'(*ap*); DNA **8**, 5'-CATCTAGTGA-3'(*p*); RNA **9**, 5'-CAUCUAGUGA-3'(*p*). Values in parentheses are T_m of parallel duplexes; nb, no binding.

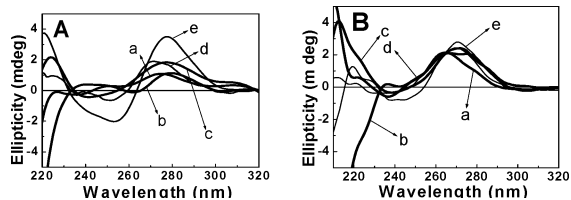


Figure 2. CD profiles: (A) PNA:DNA **6** and (B) PNA:RNA **7** duplexes. (a) **1**, (b) **2**, (c) **3**, (d) **4**, and (e) **5**. Buffer: sodium phosphate (10 mM), pH 7.0, with 100 mM NaCl and 0.1 mM EDTA.

PNA (entry 1). PNA:RNA antiparallel duplexes (**3:7**, **5:7**, entries 3 and 5) having T_m values >85 $^{\circ}\text{C}$ did not disassociate completely, whereas parallel duplexes (**3:9**, **5:9**) with a lower stability showed complete melting (Figure 1B, inset). The most important feature of the data in Table 2 is that *SR/RS*-*ch*-PNAs destabilize the DNA duplex but enormously stabilize the RNA duplex, while *SR/RS*-*cp*-PNAs remarkably stabilize both RNA and DNA complexes. In the process, the *ch*-PNAs induce remarkable differences in duplex stabilities among their DNA and RNA complexes, with $\Delta T_m(\text{RNA}/\text{DNA}) = +33$ $^{\circ}\text{C}$ for *SR*-*ch*-PNA **2** and $+50$ $^{\circ}\text{C}$ for *RS*-*ch*-PNA **3**. This is a highly significant, exceptional binding selectivity of PNA to RNA over the same DNA sequence.

Figure 2 displays the CD profiles for all *ap*-PNA:DNA and PNA:RNA duplexes. The *ch*-PNA:DNA duplexes (Figure 2A, b and c) show CD profiles similar to that of *aeg*-PNA:DNA duplex (a), with a positive band at 277 nm and a low intensity negative band at 250 nm. In comparison, the CD profiles of *cp*-PNA:DNA duplexes (Figure 2A, d and e) show a higher intensity positive band at 275 nm and a negative band (250 nm). The CD profiles of *ch/cp*-PNA:RNA duplexes (Figure 2B, b–e) consisting of a high intensity positive band (260–270 nm) and a low intensity negative band (245 nm), like that of *aeg*-PNA:RNA duplex (Figure 2B, a), suggest a high degree of helical identity among the *aeg/ch/cp*-PNA:RNA duplexes. The CD features of *cp*-PNA:DNA duplexes are akin to their RNA complexes, implying that *cp*-PNA:DNA duplexes can adopt a more RNA-like structure with higher stability. The larger elliptical intensities seen in *ch/cp*-PNA:RNA duplexes are indicative of a better base stacking in these compared to *ch*-PNA:DNA duplexes.

Classical *aeg*-PNA being flexible can easily attain competent conformation to hybridize with DNA as well as RNA and, hence, has near equal binding affinity to both as measured by T_m . In (*SR/RS*)-*ch*-PNA, the monomer dihedral angle β matches the range (65°) found in the PNA:RNA duplex, rather than that in PNA:DNA duplex (141°). The derived *ch*-PNAs thus exhibit a higher affinity to RNA and destabilize the complex with DNA. (*SR/RS*)-*cp*-PNA with a lower dihedral angle β (25°) binds to both RNA and DNA with a higher avidity compared to those of *aeg*-PNA and *ch*-PNA, but lack the differentiating ability of *ch*-PNA. In the five-membered

rings of *cp*-PNA and DNA, the flexible ring puckering allows better torsional adjustments to attain hybridization competent conformation. The inherently rigid six-membered ring of *ch*-PNAs forbids such structural adjustments and consequently destabilizes its binding with DNA. Further, in *ch/cp*-PNAs, the favorable conformational features of the monomer seem to be cooperatively transmitted to the oligomer level even in a chimera. Thus, the rigid (*SR/RS*)-*ch*-PNA imparts an unparalleled selectivity for binding to RNA over DNA for duplex formation (>30 – 50 $^{\circ}\text{C}$), while the flexible (*SR/RS*)-*cp*-PNA binds both DNA and RNA with high affinity, but without any selectivity.

SR-*ch*-PNA binds RNA only in antiparallel mode but not in parallel mode. The analogous *RS* isomer formed parallel RNA hybrid with a T_m lower than that of the antiparallel hybrid. Such orientation effects were marginal for *cp*-PNA hybrids. The differences in binding of parallel and antiparallel hybrids suggest that conformational preorganization through the dihedral angle β , the cis geometry, and the chirality (*SR/RS*) act together to impart RNA binding selectivity for *ch*-PNA.

The present results for the strong preference of (*SR/RS*)-*ch*-PNA to bind RNA is in consonance with our strategy of adjusting dihedral angle β through chemical modifications to achieve structure-based selectivity in PNAs.^{4b} This is perhaps the first report of any PNA analogue that overwhelmingly discriminates identical DNA/RNA sequences with a strong preference for RNA. The modification or conjugation of PNAs with cell-penetrating ligands may also prompt their nuclear entry, and induction of binding specificity for the DNA/RNA complement would be highly desirable to improve the selectivity and efficiency of nucleic acid-based drugs.¹³ Further studies to examine the sequence dependence on RNA/DNA discrimination properties and control of binding selectivity are in progress.

Acknowledgment. T.G. thanks CSIR, Delhi, for a fellowship.

Supporting Information Available: The synthesis details, HPLC, and mass spectra of *ch/cp*-PNAs, UV, and CD data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

References

- (1) Nielsen, P. E.; Egholm, M.; Berg, R. H.; Buchardt, O. *Science* **1991**, *254*, 1497–1500.
- (2) Egholm, M.; Buchardt, O.; Christensen, L.; Behrens, C.; Freier, S. M.; Driver, D. A.; Berg, R. H.; Kim, S. K.; Norden, B.; Nielsen, P. E. *Nature* **1993**, *365*, 566–568.
- (3) (a) Good, L.; Nielsen, P. E. *Antisense Nucleic Acid Drug Dev.* **1997**, *7*, 431–437. (b) Braasch, D. A.; Corey, D. R. *Biochemistry* **2002**, *41*, 4503–4510.
- (4) (a) Ganesh, K. N.; Nielsen, P. E. *Curr. Org. Chem.* **2000**, *4*, 1931–1943. (b) Kumar, V. A. *Eur. J. Org. Chem.* **2002**, 2021–2032.
- (5) Betts, L.; Josey, J. A.; Veal, M.; Jordan, S. R. *Science* **1995**, *270*, 1838–1841.
- (6) Brown, S. C.; Thomson, S. A.; Veal, J. M.; Davis, D. J. *Science* **1994**, *265*, 777–780.
- (7) Ericksson, M.; Nielsen, P. E. *Nat. Struct. Biol.* **1996**, *3*, 410–413.
- (8) Lagriffou, P.; Witteng, P.; Ericksson, M.; Jensen, K. K.; Norden, B.; Buchardt, O.; Nielsen, P. E. *Chem.–Eur. J.* **1997**, *3*, 912–919.
- (9) (a) Myers, M. C.; Witschi, M. A.; Larionova, N. V.; Franck, J. M.; Haynes, R. D.; Hara, T.; Grakowski, A.; Appella, D. H. *Org. Lett.* **2003**, *5*, 2695–2698. (b) Pokorski, K. P.; Witschi, M. A.; Bethani, L. P.; Appella, D. H. *J. Am. Chem. Soc.* **2004**, *126*, 15067–15073.
- (10) (a) Govindaraju, T.; Gonnade, R. G.; Bhadbhade, M. M.; Kumar, V. A.; Ganesh, K. N. *Org. Lett.* **2003**, *5*, 3013–3016. (b) Govindaraju, T.; Kumar, V. A.; Ganesh, K. N. *J. Org. Chem.* **2004**, *69*, 1858–1865.
- (11) Lambert, J. B.; Papay, J. J.; Khan, S. A.; Kappauf, K. A.; Magyar, E. S. *J. Am. Chem. Soc.* **1974**, *96*, 6112–6118.
- (12) (a) Govindaraju, T.; Kumar, V. A.; Ganesh, K. N. *Chem. Commun.* **2004**, 860–861. (b) Govindaraju, T.; Kumar, V. A.; Ganesh, K. N. *J. Org. Chem.* **2004**, *69*, 5725–5734.
- (13) (a) Hunziker, J. *Chimia* **2001**, *55*, 1038–1041. (b) Liu, Y.; Braasch, D. A.; Corey, D. R. *Biochemistry* **2004**, *43*, 1921–1927.

JA044142V